CHEMISTRY LETTERS, pp. 1317-1318, 1977. Published by the Chemical Society of Japan

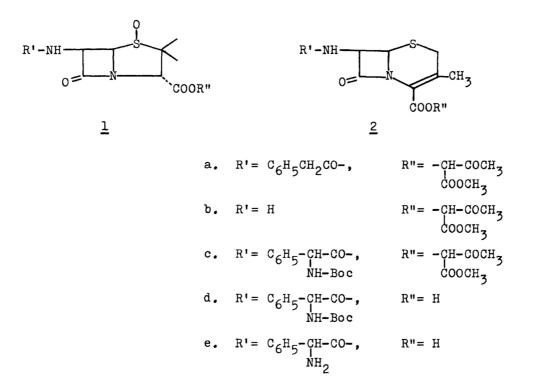
 α -HALO- β -DICARBONYL COMPOUNDS. NOVEL CARBOXYL PROTECTING REAGENTS. THE APPLICATION TO THE SYNTHESIS OF CEPHALEXIN FROM PENICILLIN

Hitoshi IKEDA, Hajime NITTA, Mariko HATAMURA, Minoru HATANAKA, and Toshiyasu ISHIMARU^{*} The Institute of Scientific and Industrial Research, Osaka University, Yamadakami, Suita, Osaka 565

 α -Halo- β -dicarbonyl compounds are successfully used as the carboxyl protective reagents in the synthesis of cephalexin from penicillin.

Our preceding report¹⁾has described that α -halo- β -dicarbonyl compounds are useful reagents for the protection of carboxylic acids, and that the regeneration of the free acids from the protected acids is carried out under mild conditions by nitrosation using sodium nitrite or alkyl nitrite in aqueous acetone. Now, we wish to report that these protective groups are applicable to the wellknown thermal rearrangement of penicillin sulfoxide into deacetoxycephlosporin²⁾. In this rearrangement, the 3-carboxyl group of penicillin sulfoxide must be protected in order to avoid the accompanying decarboxylation^{2a)}. Several esters have been used for this purpose³⁾. Our new protective groups provided a convenient process for the synthesis of cephalexin from penicillin.

Heating 1-methoxycarbonyl-2-oxopropyl 6-phenylacetamidopenicillanate 1-oxide $(\underline{1a})$ for 5 h in boiling dioxane in the presence of pyridinium dichloromethane-phosphate⁴⁾ gave 1-methoxycarbonyl-2-oxopropyl 7-phenylacetamido-3-deacetoxy-cephalosporanate ($\underline{2a}$) in 81% yield: mp 149.5°C (decomp.); IR (Nujol) 1760 cm⁻¹ (β -lactam); Calcd. for $C_{21}H_{22}N_2O_7S$: C, 56.49; H, 4.97; N, 6.27%. Found: C, 56.23; H, 4.98; N, 6.48%. Subsequently, the phenylacetyl group was removed from $\underline{2a}$ in a usual way; $\underline{2a}$ was first treated with phosphorus pentachloride in methylene chloride containing pyridine at 5-10°C for 1.5 h, following with an excess of methanol at -20°C, and then with ice water for 1 h, to provide $\underline{2b}$, which was not characterized. The crude $\underline{2b}$ was then coupled with D- α -t-butoxycarbonylamino phenylacetic acid via the mixed anhydride with ethyl chloroformate to give



<u>2c</u> (in 74% yield from <u>2a</u>): mp 145-147°C (decomp.). Treatment of <u>2c</u> with 5 equivalents of sodium nitrite in aqueous acetone containing 5 equivalents of acetic acid at 35°C for 30 min gave <u>2d</u> in 74% yield; mp 135°C (decomp.). Subsequent removal of the N-t-butoxycarbonyl group by treatment with p-toluenesulfonic acid monohydrate in acetonitrile afforded cephalexin (<u>2e</u>) in 93% yield.

References

- 1) T.Ishimaru, H.Ikeda, M.Hatamura, H.Nitta, and M.Hatanaka, Chem. Lett., preceding paper.
- 2) a) R.B.Morin, B.G.Jackson, R.A.Mueller, E.R.Lavagnino, W.B.Scanlon, and S.L.Andrews, J. Am. Chem. Soc., <u>85</u>, 1896 (1963); ibid., <u>91</u>, 1401 (1969).
 b) R.R.Chauvette, P.A.Pennington, C.W.Ryan, R.D.Cooper, R.L.Jose, I.G.Wright, E.M.Van Heyningen, and G.W.Huffman, J. Org. Chem., <u>36</u>, 1259 (1971).
- 3) E.H.Flynn, Ed., " Cephalosporins and Penicillins ", Academic Press, New York and London, Chapter 5 (1972).
- 4) Glaxo Laboratories, Germany Patent 2,011,376 (1970); Chem. Abstr., <u>74</u>, 13,172k (1971).
 - * To whom correspondences should be addressed.

(Received August 16, 1977)